



NEET - MDS

MASTERS OF DENTAL SURGERY

BY NBE

NATIONAL ELIGIBILITY CUM
ENTRANCE TEST

Volume - 5

Dental Materials, Oral Pathology, Oral
Radiology & Medicine, Pedodontics



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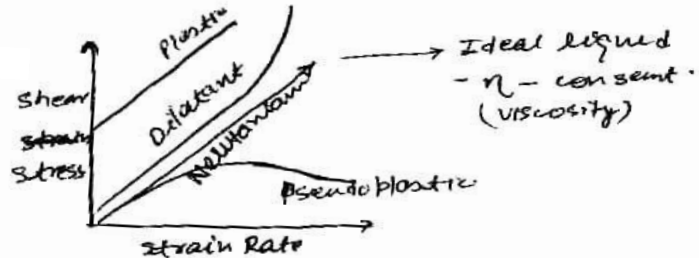
Dental Material

Physical & Mechanical Properties :-

Rheology :- Study of deformation/flow of matter.

- Viscosity \rightarrow the resistance of material to flow.

e.g cement
Restoration.



1) Newtonian liquid :- Ideal liquid

- Viscosity constant.
- Shear stress \propto strain rate.

2) Pseudoplastic :-

Force $\uparrow \rightarrow$ Viscosity $\downarrow \rightarrow$ constant. (thin)

3) Dilatant :-

Force $\uparrow \rightarrow \eta \uparrow$ (thicker)

4) Plastic :-

- Initial viscosity \uparrow ve.
- Certain amount of stress then flowing.

Thixotropic :- Force $\uparrow \rightarrow \eta \downarrow$

- Time depended
- Repeated application of pressure.

ex pop

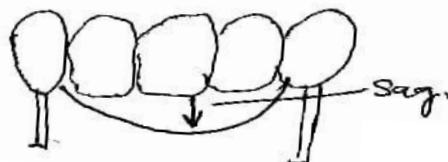
- prophylactic past.

Creep, flow, sag :-

Creep :- time depended plastic strain under static load.

Sag :- seen in long span FPD, occur porcelain firing temp.

- occur because of prostheses own weight.



Flow :- It is use ^{to denote.} rheology of a amorphous material. ex wax.

Colore :- visible range 400-700 nm
 $< 400 \rightarrow$ UV
 $> 700 \rightarrow$ IR
 cells response \rightarrow cone. \rightarrow 500 nm (green)

Hue, value, chroma :-

Hue :- Dominate colour of an object.

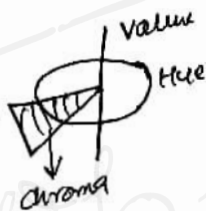
Value :- Relative brightness.

Chroma :- degree of saturation (intensity)

Munsell colour chart :- Go around periphery \rightarrow Got hue.

Go verticle $\downarrow \uparrow \rightarrow$ Got value.

Go centre \rightarrow periphery \rightarrow Got chroma



- Munsell colour system describe \rightarrow \rightarrow quantity

- LAB chart.

L - value

A - Red green axis.

B - blue yellow axes.

Bezold Brucke effect :- Brightness changes the colour of object appears to change.

Metamerism :- When sources changes \rightarrow colour changes.

Fluorescence :- Object absorbed near UV light 300-400nm when source has been remove it emit back light back 400-450nm

Thermal Properties :-

Thermal conductivity :- ^{Denote} How well heat is transmitted
 Unit \rightarrow Watt/mK \leftarrow measured at steady Rate.

Highest Pure gold $\rightarrow 297 \text{ W m}^{-1} \text{ K}^{-1}$

amalgam $\rightarrow 226 \text{ "}$

min. GIC $0.51 - 0.72 \text{ W m}^{-1} \text{ K}^{-1}$

Thermal Diffusivity :- The Rate at which object ~~the~~ attain equilibrium.

$$k = \sqrt{\kappa \cdot d} \propto \frac{1}{\text{Thermal insulation}}$$

- measured at dynamic rate.

- unit $\text{cm}^2 \text{ s}^{-1}$

Coefficient of thermal expansion :- $\frac{\Delta L}{L}$ (change in length) when

Temp $\uparrow 1^\circ \text{C}$.

- Highest $\alpha = 400$. (Dm lay wax)

- min. $\alpha = 6.6$ (Aluminous porcelain)

$$\frac{\alpha_{\text{metal}}}{\alpha_{\text{top surface}}} = 1 \quad \text{Ideal condition.}$$

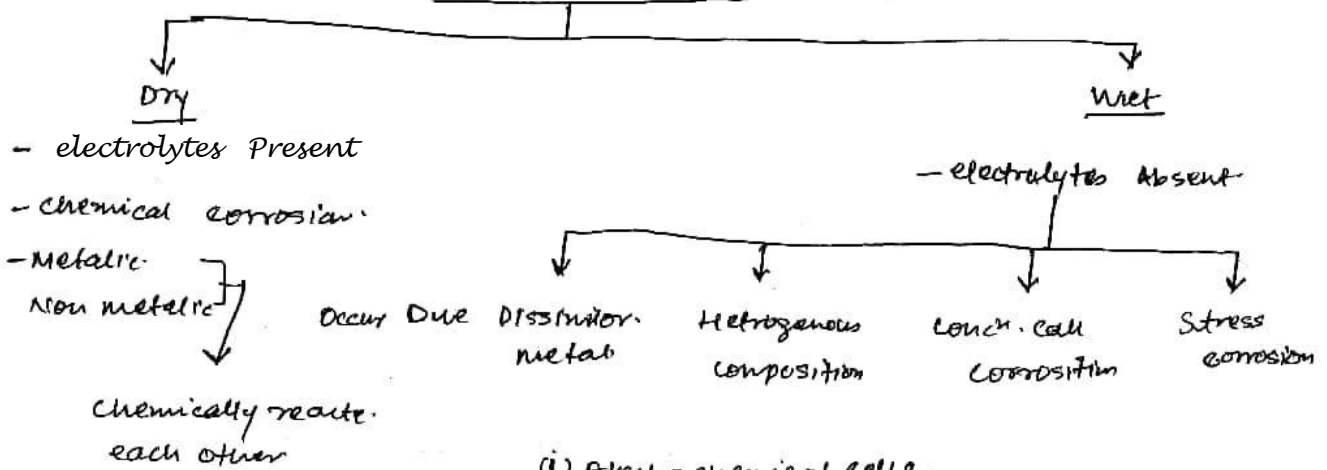
Tarnish & corrosion :- (Electrochemical Properties) :-

Corrosion :- It is electrochemical phenomenon in which the metal is attacked by some external agent resulting into complete/Partial dissolution of metal.

Tarnish :- It is surface discoloration - It does not complete/Partial dissolution of metal

- It is mainly because of Oxides, Sulphides, Chlorides.

Types of Corrosion



(i) Electrochemical cell :-

Anode. - oxidation → more reactive
- corrosion

Cathode → Reduction → Less reactive

Electrolytes → medium of ion

ion Anode → cathode

ex. Dissimilar metal (Galvanic shock)

2) Heterogeneous composition :- eg Eutectic alloys

✓ 2 or more alloys

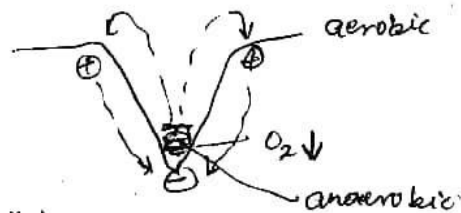
✓ Less noble alloy → corrosion

3) Stress concn :- some area deformed

↓
↑ internal energy

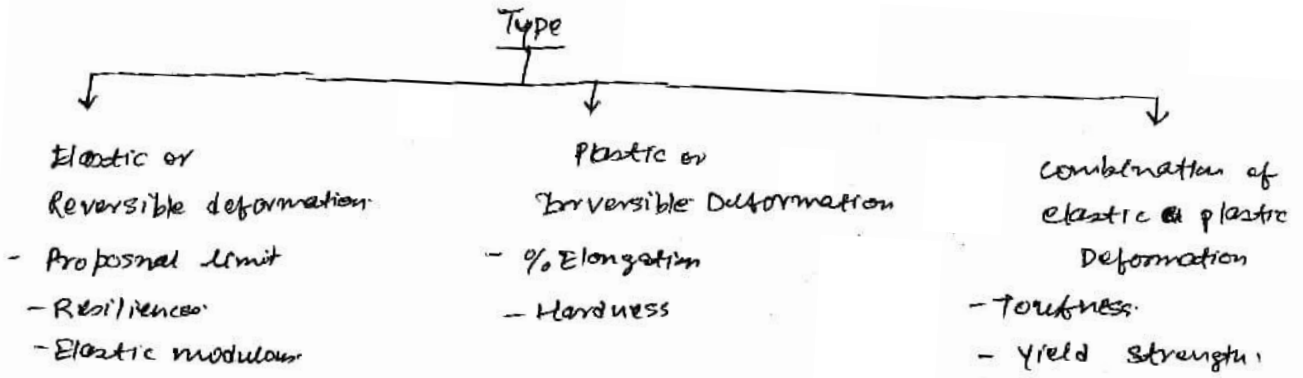
↓
Corrosion

4) Concn cell corrosion :- r/a pitting corrosion



✓ O_2 concn ↓

Mechanical Properties :-



Stress :- F/A

Strain :- $\frac{\Delta L}{L}$

Types of strain stress

1) Compression strain stress



2) Tensile



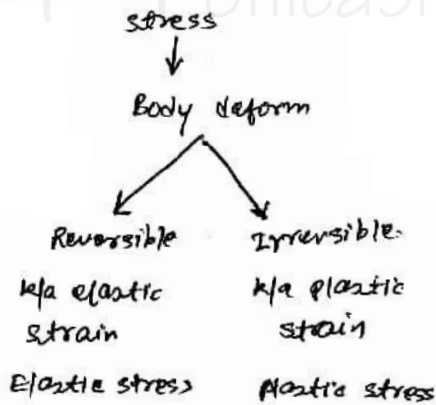
3) Shear



4) Torsion



5) Bending (flexural)

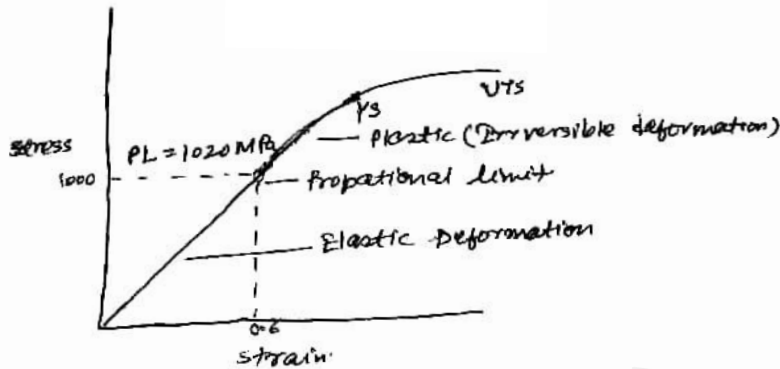


Elastic Properties :- measure Elastic & Plastic strain

- Elastic modulus
- Dynamic modulus
- shear modulus
- Plaxibility
- Resilience
- Poisson Ratio

Elastic modulus :- k/a young modulus / modulus of elasticity.

- Describe Relative Rigidity / stiffness of a material.
- Measure by slope of stress - strain graph but below proportional limit.



- \uparrow slope $\rightarrow \uparrow$ elastic modulus $\rightarrow \uparrow$ stiff / Rigid

- constant

- Independent of ductility

- Not measure of strength

- For a given stress \rightarrow lower strain produced $\rightarrow \uparrow$ Elastic modulus

- E. modulus = $\frac{\text{Stress}}{\text{Strain}}$ Stress value < Proportional limit (PL)

Dynamic young's modulus :- Elastic modulus can be measure by

- Static Tech.

- Dynamic Tech. \rightarrow use velocity of sound

- Density of material.

Shear modulus (G) = 38% E. modulus.

Flexibility :- small stress \rightarrow resultant deformation is \uparrow (larger strain is produced)

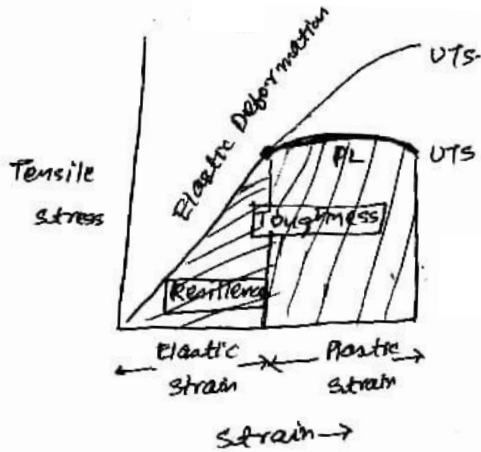
\downarrow
Flexible

Resilience :- Interatomic spacing $\uparrow \rightarrow$ Internal energy \uparrow

If stress < PL this energy called as resilience.

- The area under elastic region of stress - strain graph.

- Total area under stress strain graph is Toughness.



- Till stress < PL energy absorbed / area vol.

- Poisson's Ratio :-

$$\nu = - \frac{E_x \text{ (strain produce x-axis)}}{E_z \text{ (strain produce z-axis)}}$$

Strength :- Stress value

fracture.
Produce certain amount of deformation

- 1) Proportional limit :- stress value till which stress & strain.
- stress above prop. limit → plastic deformation.
 - stress < prop limit → elastic deformation.
 - stress > prop limit → stress & strain

2) Elastic limit :- It is max. stress which can be subjected to a body such that it returns to its origin dimension.

3) Yield strength :- where prop. limit can not be determinate accurately.

- stress value reqd to produce a particular amount of strain (0.1%, 0.2%)

↓
(% offset)

Diametric tensile strength :- for brittle material

e.g. Brazilian test.

↓
measure the strength

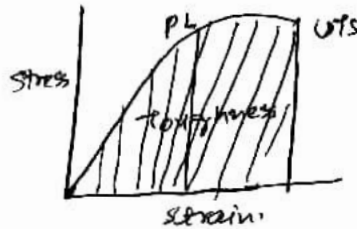
kPa Transverse strength

- modulus of rupture.

Flexural strength :-

Impact strength:- Energy require to fracture material under impact force.
 It is Charpy type & Izod type.

Toughness:-



fracture toughness:- For brittle material, strength value are not applicable

- They can fracture below $<$ ultimate strength value.
- because of fatigue failure.
- fracture toughness: (critical stress intensity)
 \downarrow
 stress.

Brittleness:- It is inability of material to resist plastic deformation before fracture of material.



Ductility:-

Test for Ductility

- 1) % Elongation \rightarrow m/c type.
- 2) cold bend test
- 3) Reducⁿ area of tensile strength specimen.

- ductility measure grain elongation.

Hardness:- It is ability of material Resist scratching/indentation of a material.

Brinell tests:- It is use for Metallic a Non Metallic
 - It is related to pro. limit a ultimate tensile strength (UTS) of an alloy.
 - It uses a hardest steel ball.
 - $BHN = \frac{\text{Load}}{\text{Projected area}}$

Rockwell tests:- use conical diamond.

- use for Metallic a Non Metallic
 both brinell a Rockwell test cannot use for brittle material.

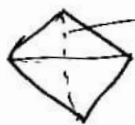
Vickers tests:- - sq. base pyramid.

$$= \frac{\text{load}}{\text{area of indentation}}$$

- use for Dental casting gold alloys.
- use for tooth structure, enamel, brittle material.

Knoop Hardness test:- Diamond with tooth

- Rhombic in outline.

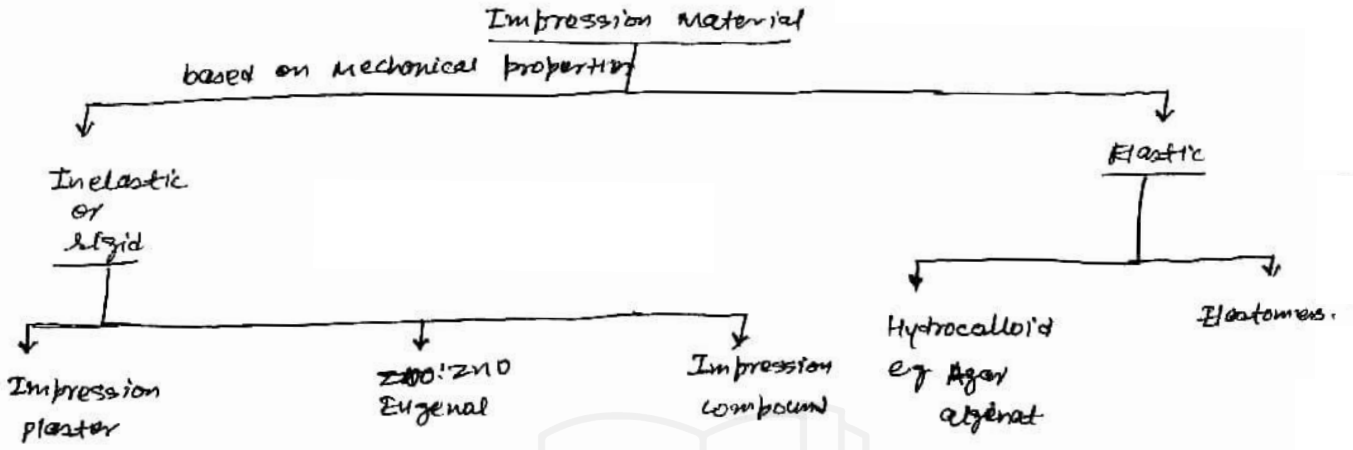


Elastic recovery - shorter diagonal.

- It is independent of ductility.
- use for ductile material, Gold, Porcelain.
- Knoop a Vickers test known as microhardness test.
- Brinell test, Rockwell test - macrohardness test.
- share a ball test → Rubbers, plastic

Impression Materials

1) Classification



2) Classification



Elastomers :-
 - synthetic polymeric
 - chemically cure

Type

- ✓ polysulphid
- ✓ Polyethers
- ✓ silicons

Addition silicon
 Condensation ✓

Composition :-

1) Polysulphid :-
 k/a Mercaptan/Thioal
 (-SH gp)

Base Post :-

- PS (Polysulphid) polymer
- plasticizer - Dimethyl phthalate
- fillers -> TiO₂ / Lithopone
- SB -> accelerators

setting reaction H₂O is a by product

- Catalyst :- PbO_2 (cross linking agent) \rightarrow brown colour.

- Plastisizer - dibutyl phthalate, (DBE)

- Retarder \rightarrow Oxalic / stannic acid.

2) Condensation silicones :- like conventional silicon.

Composition :-

Base

Catalist

α - ω Hydroxyl Terminated polydimethyl siloxane.

- Tri-Tetra functional alkyl silicate.

Setting rxn :- Ethyl alcohol is a by product

- stannous octoate - octoate catalyst.

\downarrow
Reason for low dimensional stability.

3) Addition silicones :- like polyvinyl siloxane, vinyl polysiloxane.

Composition :-

Base part

catalist

- Poly methyl Hydrosiloxane (PMHS)
- ~~Di~~
catal

- Divinyl polysiloxane
- Platinum salt \downarrow
- palladium \rightarrow Scavenger

Setting rxn :- NO by product

- Excellent Dimensional stability.

- Unreacted PMHS + moisture \rightarrow H_2 gas.

- Silicones are Hydrophobic \rightarrow Non ionic surfactant make it Hydrophilic

- ~~SS~~ sulphur \rightarrow inhibit setting rxn

- use Nitrill gloves

- Not latex gloves are use.

4) Polyether:- composition:-

Base part:- polyether polymer

- Plastisizer - DBP

- Filler - Colloidal silica

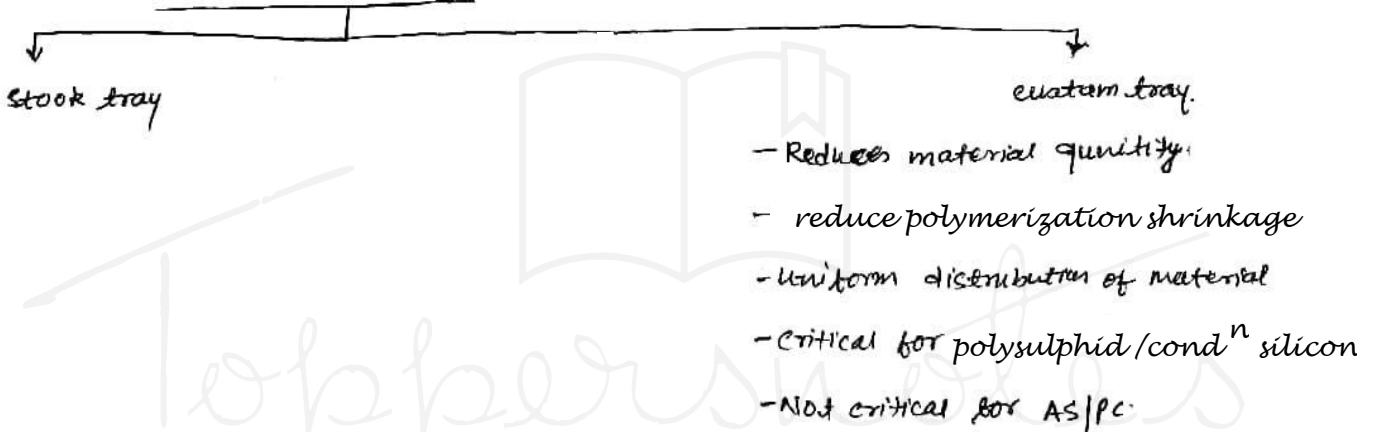
catalyst part:-

- Alky. aromatic sulphonates

- polyether has aziridine rings.

- Polymerization take place via imine group.

Impression trays



Tray adhesive:-

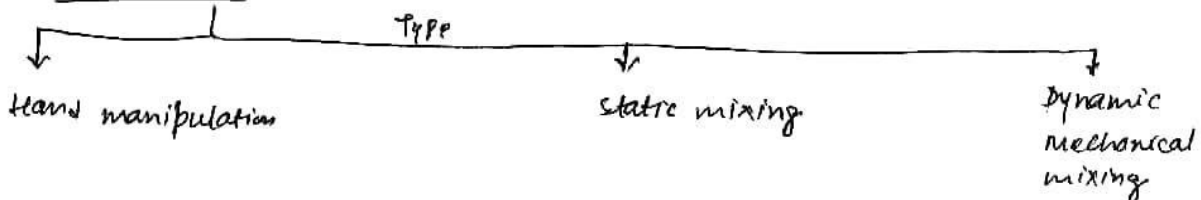
PS. → Butyl Rubber

- Styrene/acrylonitril.

These are dissolved in solvent chloroform.

- when use silicones - Ethyl silicate

Manipulation:-



Removal of impressions:- 10 mint \rightarrow setting time.

polyether \rightarrow Hard to remove.

Poly sulphid \rightarrow more flexible (easy remove)

all elastomers \rightarrow viscoelastic material.

\rightarrow remove quick snap (no rocking mov.)

Stone cast & dies:-

Casts

Silicones \rightarrow Hydrophobic

Epoxy resin \rightarrow Hydrophobic

Gypsum \rightarrow voids

- Hydrophilic

- wetting agent / surfactant

- Pe/PS \rightarrow Hydrophilic

- Pe/AS \rightarrow permit multiple cast pouring

Working time / setting type:-
(WT) (ST)

start of mixing \rightarrow just before elastic properties develop.

SP \rightarrow start of mixing \rightarrow sufficiently cured (set) to be removed without disintegration.

- Filler \uparrow \rightarrow WT / ST \downarrow

- Temp \uparrow \rightarrow WT / ST \downarrow

\uparrow catalyst / ~~base~~ \rightarrow WT / ST \uparrow \rightarrow base : catalyst ratio fixed. If u change

\downarrow
physical properties \downarrow

Rheological properties:- - mouth - introduce as a viscous past.

- shear thinning

Elasticity & Viscosity:-

under cates⁺ →

Add. silicons (best)

- PS.

- Highest elastic recovery → AS (Addition sililones)

- Highest deformation on removable → PS.

↓
PS > CS > Pe > Ad's

- maxwell - Voigtlet

- Use snape.

- Easest to Remove → Polyscelphid

- Should be used → AS

Stiffness:-

Pe > AS > CS > PS

↳ Least stiff

↳ most flexible.

Tear strength:-

T.S → AS > CS > Pe > PS

Dimensional Stability:-

CS < PS < Pe < AS.

Disinfection:-

Pe - enain compound.

- iodophores.

CS/AS/PS → enain compound

- iodophores

- Gluteraldehydes

- Phenolic compound

Pe < 10 mint (emersion time)

↑ emersion AS → ↓ Hydrophobic.

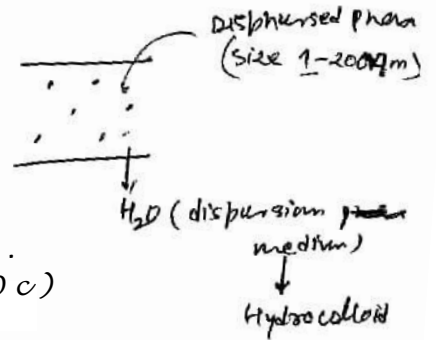
Biocompatibility:-

- least cytotoxic → ps.
- most " → pe.

Hydro colloid :- 4th state of matter.

Setting rxn:-

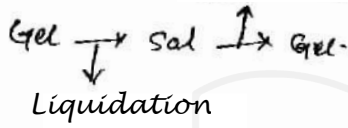
- asso. with formation of micelles. a brush heap structure.



Gelation Temp. (37° - 50° c)

Agar :-

Setting rxn:-



Temp
(72° - 100° c)

- gel gelation tem < oral cavity
↓
make canal impression

Component

function

Composition (%)

<u>Component</u>	<u>function</u>	<u>Composition (%)</u>
Agar	Brush Heap structure	8-15%
Borate.	strength	0.2 - 0.5%
Potassium sulphates	Gypsum Hardner	0.5 - 1.0% 1-2%
Filler (silica, clay)	control, strength, viscosity & rigidity.	0.5 - 1.0%
Rhixotropic Material	- thickner	0.3 - 0.5
Water	- Reaction medium	78%

- more active ingredient → Agar
- max quantity → water
- Thymol & glycerin → bacteriocidal.