

IES / GATE

Electrical Engineering

VOLUME-III

**Basic Electronics Engineering
Control System**

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ELECTRONIC DEVICES & CIRCUITS

1. Integrated Electronics
by Millman Halkias
2. Micro Electronics
by Sedra & Smith.
3. Electronics devices & circuits
By Neamen.

Classification of Temperature:-

- (1) Absolute Temp $0K = -273^{\circ}C$
- (2) Room Temp $300K = 27^{\circ}C$ [for semiconductor devices by default this temp is considered]
- (3) Ambient Temp $[T_A] 290K = 17^{\circ}C$ [for communication system Ambient temp^s (surrounding temp^s) took as default temp^s].

$$\text{Temp in } ^{\circ}C = \text{Temp in Kelvin} - 273$$

$$^{\circ}K = K$$

Thermal voltage :- (V_T) :-

* "Volt equivalent of temperature"

$$V_T = \frac{KT}{q} \text{ volt.}$$

* (Boltzman Constant):

$$K = 1.381 \times 10^{-23} \text{ J/K.}$$

$$k = 8.62 \times 10^{-5} \text{ eV/K.}$$

T = Temperature in K.
 q = magnitude of charge
 $= 1.6023 \times 10^{-19} \text{ coulomb}$
 $K = 1.381 \times 10^{-23} \text{ J/K.}$

$$V_T = \frac{T}{11600} \text{ volt.}$$

$$V_T \propto T$$

If $T = 0K \quad V_T = 0$

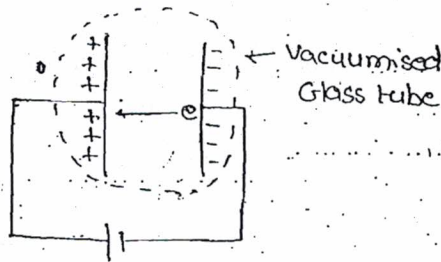
If $T = 300K \quad V_T = \frac{300}{11600} = 0.02568 \text{ volt}$
 $\approx 26 \text{ mV}$

* The standard room temperature corresponds to a voltage of 26.

For a large variation in temperature we get a minute variation in thermal voltage.

Electron volt (eV):-

- * It is the practical unit of energy in electronics.
- * 1 eV is defined as the energy gain by the electron in moving through a potential difference of 1 volt.



$$\epsilon_r(\text{air}) = 1.01 \approx 1$$

$$\epsilon_r(\text{Vacuum}) = 1$$

$$1 \text{ eV} = |q| \times \text{P.D.}$$

$$= 1.6 \times 10^{-19} \times 1 \text{ Volt}$$

$$= 1.6 \times 10^{-19} \text{ Coulomb-Volt or Joule.}$$

$$1 \text{ eV} = 1.6 \times 10^{-19} \text{ Joule.}$$

* Electron volt indicates the kinetic energy gained by the electron or potential energy lost by the electron.

$$\text{K.E.} = \frac{1}{2} m v^2$$

↳ mass of e⁻ (9.1 × 10⁻³¹ k.g.)

$$\text{P.E.} = q \cdot V$$

$$\text{K.E. gained} = \text{P.E. loss}$$

$$\frac{1}{2} m v^2 = qV$$

$$\text{Velocity of } e^- \quad v = \sqrt{\frac{2qV}{m}} \text{ meter/sec}$$

Electric Field Intensity (E or E):-

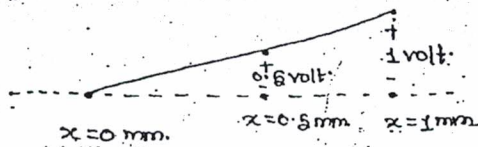
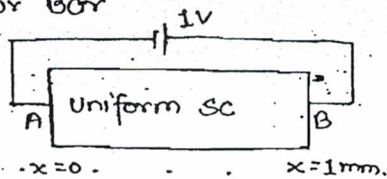
* Normally called as field intensity, field gradient or field.

By Definition

$$E = -\frac{dV}{dx} \text{ volt/meter}$$

$$|E| = \frac{|\text{Voltage existing}|}{\text{spacing or distance}}$$

Ex. Considering a uniform semiconductor bar



Magnitude of field intensity at end B.

$$|E|_B = \frac{|V_B|}{x_B} = \frac{1 \text{ volt}}{1 \times 10^{-3} \text{ m}} = 10^3 \text{ v/m.}$$

Magnitude of field at centre of bar.

$$|E|_C = \frac{|V_C|}{x_C} = \frac{0.5}{0.5 \times 10^{-3}} = 10^3 \text{ v/m.}$$

$$|E|_A = \frac{|V_A|}{x_A} = \frac{0}{0} = \text{undefined.}$$

Mobility of Charge Carrier (μ):

Mobility denotes the current carrying capacity or how fast the charge carrier is moving from one place to another place.

Mobility is defined as drift velocity to field intensity.

$$\mu = \frac{V_d}{E} \quad \text{m}^2/\text{V}\cdot\text{s} \text{ or } \text{cm}^2/\text{V}\cdot\text{sec}$$

$$\mu = \frac{\text{Drift Velocity}}{\text{field intensity}}$$

e⁻ mobility (μ_n or μ_e).

hole mobility (μ_p or μ_h)

μ	Ge	Si
μ_e	3800 cm ² /v-sec	1300 cm ² /v-sec
μ_p	1800 cm ² /v-sec	500 cm ² /v-sec

$$\frac{\mu_n}{\mu_p} = 2.1 \quad (\text{for Ge})$$

$$\frac{\mu_n}{\mu_p} = 2.6 \quad (\text{for Si})$$

Electron mobility is always greater than hole mobility.

Electron can travel faster & also contribute more current than hole.

Ge → Higher Conductivity (Due to larger mobilities)
 Relatively more suitable for high frequency application [due to large B.W]

Si → Relatively more suitable for swi applications (Better thermal stability).
 Best suit for high power applie (Natural Property).

Mobility of charge carrier always decreases with the temperature.

As temperature increases atoms of the material will vibrate & due to this thermal vibration or the agitation the mobility of charge carrier

$$\mu \propto T^{-m}$$

where m is constant and is given

as

For Si

For Ge

$$m = 2.5 \text{ for } e^-$$

$$m = 1.66 \text{ for } e^-$$

$$= 2.7 \text{ for hole.}$$

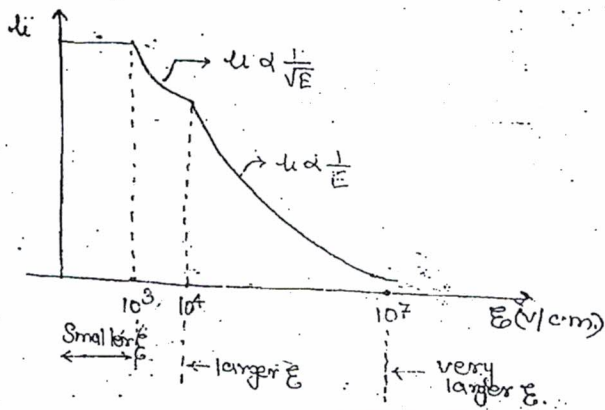
$$= 2.33 \text{ for hole.}$$

Mobility decreases with the temperature

as a non linear variation.

μ vs E curve for a S.C. :-

(Experimentally plotted).



Question: when smaller field intensities

are applied to the semiconductor,

(a) mobility of charge carriers remain constant.

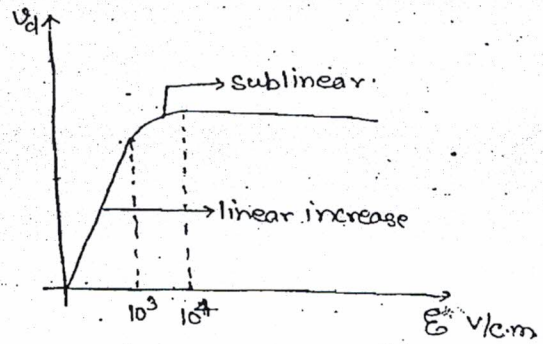
(b) drift velocity linearly increases with the field intensity.

Question: when larger field intensities are applied to the semiconductor

(a) mobility of the charge carrier decreases.

(b) drift velocity enters into saturation.

v_d vs E curve for S.C. :-



Question: In a semiconductor when

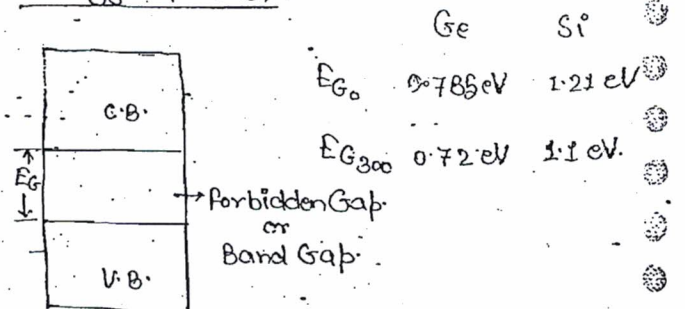
field intensity are increasing the drift velocity μ -

(a) linearly increases.

(b) sublinearly increases.

& enters into saturation with larger fields applied.

Energy Gap (E_g):-



In semiconductor, E_g (or) energy gap decreases with temperature.

$$E_g \propto \frac{1}{\text{Temp}(T)}$$

$$E_g(T) = E_{G_0} - \beta_0 T \text{ eV.}$$

β_0 = material constant (eV/°K)

For Si $\beta_0 = 3.6 \times 10^{-4} \text{ eV/°K}$

$$E_g(T) = 1.21 - 3.6 \times 10^{-4} T$$

for Ge $\beta = 2.33 \times 10^{-4}$

So $E_G(T) = 0.785 - 2.33 \times 10^{-4} T$

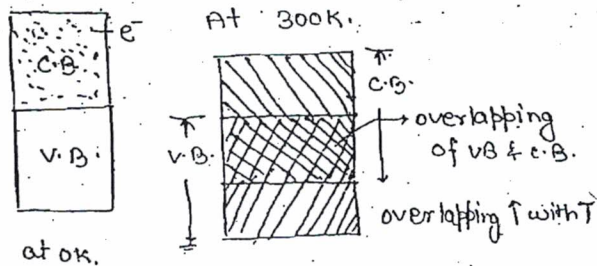
Classification of Elements into conductors, semiconductors, & insulators:-

Conductors or Metals:-

- * Very good conductors of currents.
- * Valence electrons are ≤ 3
- * All metals are unipolar.
- * I, II & III group of periodic table (metallic or non metallic).
- * In metal free e^- concentration is very high.

Concentration of $e^- (n) = 10^{28} / m^3$

- * metallic bonding is present
- * Energy gap $E_G = 0$ at 0K.



- * In metals free electrons are available even at 0K.
- * In metals free electron concentration is independent of temperature.
- * Because of overlapping a valance band & conduction band metal will exhibit PTC of resistance.

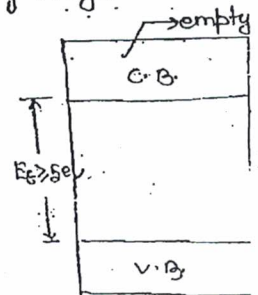
* In metals there is only a dir current.

E.g. → Gold, Silver, Tungsten, Platinum, minimum, Copper, Tin, Iron, Uranium.

Insulators:-

- * Bad conductors of current.
- * Valency electrons are 7 or 8.
- * Ionic bonding
- * NTC (Negative Temperature Coeffici of resistance.
- * Energy gap is very large.

- * $E_G \geq 5 \text{ eV}$.
- * Conductivity is zero for ideal insulators.
- * $\rho =$ negligible for practical insulators.



- * In insulators E_G (Energy Gap) slit decreases with the temperature. ↑
- * Insulators are subjected to break-down.

E.g. Diamond, SiO_2 , Air, Mica, ceramic, porcelain, Bakelite, Paper, Rubber, Col: PVC, Leather, wood, glass, Plastic, fiber.

SemiConductor's (S.C):-

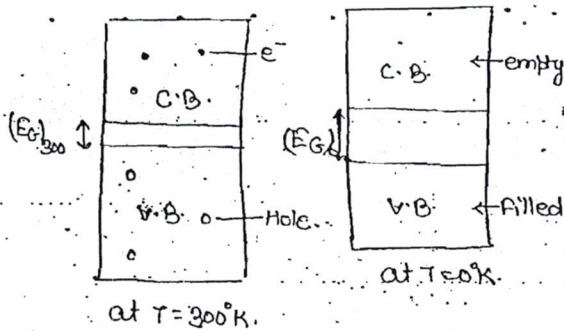
- * Semi-conductors are elements wh conductivity lies between the conductivity of an insulator & conductivity of a conductor.
- * Valance electrons are 4. (IV group between the con)

2. Covalent Bonding.

3. All semiconductors are insulators at 0°K.

4. Energy gap is small (0.7 eV to 1.5 eV) or ≈ 1 eV.

Energy Band Diagram at 0°K.



5. At 0°K carrier concentrations are zero & therefore conductivity is 0 & semiconductor at 0°K will be working as an insulator.

6. At 300°K because of thermal energy a no. of covalent bond will be broken & equal no. of electron & hole's are created & there will be a conductivity in a semiconductor.

7. Semiconductors are Bipolar having two different type of charge carriers - electrons & holes.

8. In a semiconductor there will be a diffusion current.

9. All semiconductors are temperature sensitive.

10. In a SC energy gap decreases with temperature.

$$E_g \propto \frac{1}{\text{Temperature}}$$

E.g. - Si, Ge...

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Why carbon is not considered as a SC element?

* Carbon belongs to IV group of the periodic table but it is never considered as a semiconductor element because energy band gap is more than 1.5 eV.

It has very unreliable & unpredictable properties.

Graphite \rightarrow behaves as SC & conducts at different temperature.

Diamond \rightarrow behaves as insulator.

Einstein's Equation:-

* It was given just as a mark of respect to great physicist Einstein.

* In a SC

$$\frac{D_n}{\mu_n} = \frac{D_p}{\mu_p} = V_T$$

* It gives the relationship b/w diffusion constant, mobility & thermal voltage.

$$\frac{D_n}{\mu_n} = \frac{D_p}{\mu_p} = \frac{kT}{q} = \frac{1}{q} V_T$$

* The unit of mobility to diffusion constant is v^{-1}

* The unit for $\frac{D}{u}$ is volt.

Diffusion Constant of Charge Carriers:

e⁻ diffusion constant (D_n) = $u_n V_T$
 the hole diffusion constant (D_p) = $u_p V_T$

Unit for $D \rightarrow cm^2/sec$ or m^2/sec .

$D = uV_T$

* Diffusion constant of charge carrier decreases with the change of temp.

Diffusion constant for Ge at 300K.

$D_n \approx 99 cm^2/sec$

$D_p \approx 47 cm^2/sec$

For Si at 300K.

$D_n = 34 cm^2/sec$

$D_p = 13 cm^2/sec$

$\frac{D_n}{D_p} = \frac{u_n V_T}{u_p V_T}$

or $\frac{D_n}{D_p} = \frac{u_n}{u_p}$ or $D_n u_p = D_p u_n$

For Ge

$\frac{D_n}{D_p} = 2.1$

For Si

$\frac{D_n}{D_p} = 2.6$

* Diffusion constant is a material const associated with the property called diffusion & it can not be negative.

MASS ACTION LAW:-

$n_p = n_i^2$

mass action law states that -

"In a sc. (intrinsic or extrinsic) under

thermal equilibrium the pro of electrons & holes is also a constant & it is equal to square of intrinsic concn

* The law is particularly used extrinsic semiconductors to calculate minority carrier concentration

N-Type Semiconductor

Majority carriers are e⁻s = n ,
 Minority carriers are holes = p

$n_p = \frac{n_i^2}{n}$

P-Type semiconductor:-

Majority carriers are holes = p ,
 Minority carrier are e⁻s = n

$n_p = \frac{n_i^2}{p}$

In a pure semiconductor e⁻s & hole concentrations are respectively by adding impurity atom into the semiconductor a holes concentration are n respectively. then following relation are applicable.

$n_1 p_1 = n_2 p_2 = n_i^2$

Intrinsic Concentration (n_i):-

Intrinsic \equiv Pure

$n = p = n_i$

$n_i^2 = A_0 T^3 e^{-E_g/KT}$

$$n_i = \sqrt{A_0} \cdot T^{3/2} e^{-\frac{E_g}{2KT}}$$

A_0 = Material constant

* Intrinsic concentration is a function of temperature & energy gap.

In a sc, Intrinsic concentration (n_i^2)

(a) T^3 (b) $T^{3/2}$

In a sc, Intrinsic concentration n_i is

(a) T^3 (b) $T^{3/2}$

* Intrinsic concentration increases with the temperature as a nonlinear variation

* when compared to Silicon, germanium has larger value of n_i & this is due to smaller value of energy gap.

At room

In Ge, $n_i = 2.5 \times 10^{13} \text{ atoms/cm}^3$

Si, $n_i = 1.5 \times 10^{10} \text{ atoms/cm}^3$

Resistivity (ρ):

(Specific Resistance)

Unit for $\rho \rightarrow \Omega\text{-cm}$
or $\Omega\text{-m}$

* In metal resistivity increases with temperature.

* In semiconductor resistance decreases with temperature.

Conductivity (σ):

* The Reciprocal of resistivity $\frac{1}{\rho}$

* It denotes current carrying capacity of device or material

* It denotes current

$$\sigma = \text{Carrier Conc}^n \times q \times \text{mobility}$$

* Conductivity depends on carrier concentration, magnitude of charge & mobility.

* Conductivity variation may be due to variation in carrier concentration & variation in mobility.

$$\sigma = n q \mu_n$$

* In metals conductivity decreases with temperature.

* In metal's free electron concentration is independent of temperature so conductivity does not depend on carrier concentration.

For semiconductors (Bipolar)

$$\sigma = n q \mu_n + p q \mu_p$$

* In intrinsic semiconductor, conductivity increases with the temp.

* In intrinsic semiconductor, as temp increases mobility decreases and this will slightly reduce the conductivity & at the same time because of thermal energy a large no. of covalent bond will be broken & equal no. of e^- & holes are generated.

& this will increase the conductivity by a larger value & that's why conductivity increases with temp in intrinsic sc.

* In sc conductivity mainly depends on carrier concentration.

Current Density (J):

It is the current passing per unit area.

$$J = I/A \dots A/m^2 \text{ or } A/cm^2$$

$$J = \sigma E \quad E \rightarrow \text{electric field intensity}$$

In metals

$$\text{Current density (J)} = nq\mu E$$

In sc

$$J = nq\mu_n E + Pq\mu_p E$$

$$J = qE (\mu_n n + \mu_p p) \quad A/m^2$$

Current:-

* Rate of change of charge is known as current.

$$i = \frac{dq}{dt} \quad \text{Ampere}$$

Drift Current:

It is the flow of current through the material or device under the influence of electric field intensity

Diffusion Current & Diffusion:-

Diffusion is a natural property. Diffusion is defined as the migration of charge carriers from higher concentration to lower concentration.

or from higher density to lower density.

$$\text{Gradient (slope)} = \frac{d}{dx}$$

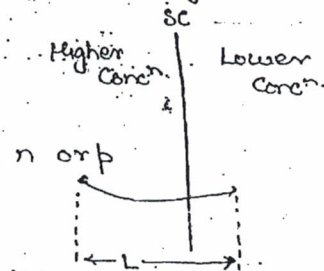
$$\frac{dn}{dx} = e^- \text{ concentration gradient}$$

$$\frac{dp}{dx} = \text{hole concentration gradient}$$

$$\text{Unit } \frac{dn}{dx} = e^- / cm^3 / cm$$

$$\text{Unit } \frac{dp}{dx} = \text{holes} / cm^3 / cm$$

* Diffusion is mainly occurred due concentration gradient.



* Diffusion current flows only semiconductor's.

* In a semiconductor diffusion is due to unequal distribution of charge carriers.

Note: In metals concⁿ of e⁻ are very high & also they exist with equal distribution & therefore there is no diffusion.

Length of Diffusion:-

$$L = \sqrt{\tau D} \quad \text{cm}$$

$$\text{but } D = \frac{1}{2} v_T$$

$$L = \sqrt{\tau \mu V_T} \quad \text{cm}$$

$\tau \rightarrow$ carrier lifetime

$\tau \rightarrow$ Average carrier life

* carrier lifetime is average lifetime

* Length of Diffusion is average length.

* Length of Diffusion depends on diffusion constant, carrier lifetime mobility of charge carriers & tempⁿ.

$$\begin{aligned} \text{e}^- \text{ Diffusion length } L_n &= \sqrt{D_n \tau_n} \text{ cm.} \\ \text{holes " " } L_p &= \sqrt{D_p \tau_p} \text{ cm.} \end{aligned}$$

$$\tau_n \neq \tau_p$$

e⁻ Diffusion current density [$J_n(\text{diff})$]:-

$$J_n(\text{diff}) = +q D_n \frac{dn}{dx} \text{ A/cm}^2$$

hole Diffusion current density [$J_p(\text{diff})$]:-

$$J_p(\text{diff}) = -q D_p \frac{dp}{dx} \text{ A/cm}^2$$

e⁻ Diffusion current $I_n(\text{diff}) = J_n(\text{diff}) \times \text{Area}$

hole Diffusion current $I_p(\text{diff}) = J_p(\text{diff}) \times \text{Area}$

* If Area is not specified in the problem then by default always consider unit cross sectional area.

* Total current density in a semiconductor

$$J = J_n + J_p$$

$$J_n = (J_n)_{\text{drift}} + (J_n)_{\text{diff}}$$

$$= nq\mu_n E + q D_n \frac{dn}{dx}$$

$$J_p = (J_p)_{\text{drift}} + (J_p)_{\text{diffusion}}$$

$$= -p q \mu_p E - q D_p \frac{dp}{dx}$$

$$J = \underbrace{q n \mu_n E}_{(J_n)_{\text{drift}}} + \underbrace{q D_n \frac{dn}{dx}}_{(J_n)_{\text{diffusion}}} + \underbrace{q p \mu_p E}_{(J_p)_{\text{drift}}} - \underbrace{q D_p \frac{dp}{dx}}_{(J_p)_{\text{diffusion}}}$$

$$J = q n \mu_n E + q p \mu_p E + 2 D_n \frac{dn}{dx} - q D_p \frac{dp}{dx}$$

* Drift current depends on carrier concentration, mobility of charge carriers & field intensity.

* Drift current mainly depends on electric field intensity.

* Diffusion current mainly depends on concentration gradient.

Operating Temperature's

↳ for Ge

-60°C to +75°C.

Maximum operating temperature = 75°C.

↳ for Si

-60°C to 175°C

Maximum operating temperature < 175°C

⇒ Normal Working temperature:-

look to 400K

Leakage Currents (I_0):-

* This is also called as reverse saturation current or minority carrier current.

* It is also called as temperature generated or thermally generated current.

↳ It never depends on applied voltage across the material.

* This current is saturated w.r.t voltage. Only when we will increase voltage the current will not vary (saturates).

* It increases with increase in temperature.

* For 1°C rise in temperature, the value of current increases by 7%.

* I_0 gets doubled by increasing 10°C of temperature in both Ge & Si.

$$I_0 = I_0 \cdot 2^{\frac{(T-T_0)}{10}}$$

	Ge	Si
I_0	4A	nA

$$I_0 \text{ of Ge} > I_0 \text{ of Si}$$

* I_0 depends on minority carriers and minority carrier concentration will be depending on temperature. Hence this current is generated

only because of temp^r. hence called thermally generated current.

* It is highly sensitive to temp^r.

* For better performance leakage currents must be smaller.

* If leakage currents are small the temperature effect on the material or device will be small & this indicates better thermal stability.

* Si is having better thermal stability than Ge.

* The greatest advantage of Si is smaller leakage currents.

Conductivity sensitivity:-

1. In intrinsic semiconductor, conductivity increases with the temperature.

for 1°C rise in Ge, σ ↑ by 6%.

for 1°C rise in Si, σ ↑ by 8%.

2. when compared to Ge, Si is having higher sensitivity to the temperature but Si is more suitable for high temperature application & this is due to the smaller leakage currents.

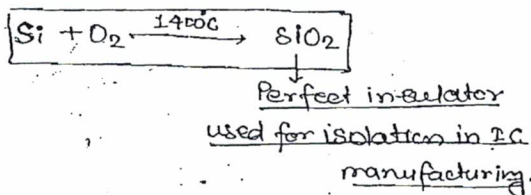
Electrical Properties of Ge & Si

↓ Properties	Ge	Si
1. Atomic Number	32	14
2. Total no. of Atoms (Density of atoms)	$4.421 \times 10^{22} / \text{cm}^3$	$5 \times 10^{22} / \text{cm}^3$
3. Intrinsic Concentration (n_i) _{300K} (atom/cm ³)	2.5×10^{13}	1.5×10^{10}
4. Intrinsic Resistivity ρ_i (300K) (Ω-cm)	45	2,30,000
5. Leakage Current (I_0)	4A	nA
6. Max ^m operating Temp ^r	75°C	175°C
7. Power handling Capability	low power	high power

* when compare to Ge, Si. is more

preferable due to

- [a] Smaller leakage Current.
 - [b] High temperature application.
 - [c] High power handling
 - [d] Si is plenty available on the surface of earth \rightarrow This is the primary reason why silicon is widely used by semiconductor's device manufacturer's.
 - [e] Si is very cheap & economical.
 - [f] Favourable property to form SiO_2
- \downarrow
- This is the main reason why Si is very fancy for IC manufacturing.



Major Disadvantage of Silicon:-

* Smaller Conductivity than Ge.

Note:- Silicon when exposed to $1400^\circ C$ will melt & we get liquid Silicon & when reacted with the oxygen we get SiO_2

• SiO_2 is a perfect insulator

SiO_2 is used to provide isolation in between the components during the IC manufacturing.

Minimum Conductivity in Semiconductor

The conductivity of a semiconductor is

$$\boxed{\sigma = nq\mu_n + pq\mu_p} \quad \text{--- (i)}$$

By mass action law we can write

$$p = \frac{n_i^2}{n} \quad \text{--- (ii)}$$

Substituting (ii) in eqⁿ (i)

$$\sigma = nq\mu_n + q\mu_p \cdot \frac{n_i^2}{n} \quad \text{--- (iii)}$$

differentiating above equation w.r.t n

$$\frac{d\sigma}{dn} = q\mu_n + q\mu_p n_i^2 \left(-\frac{1}{n^2}\right)$$

$$\frac{d^2\sigma}{dn^2} = q\mu_p n_i^2 \left(\frac{2}{n^3}\right)$$

since second derivative is positive

we get the condition for minimum conductivity by $\frac{d\sigma}{dn} = 0$

$$q\mu_n + q\mu_p n_i^2 \left(-\frac{1}{n^2}\right) = 0$$

$$\mu_n = \mu_p \cdot \left(\frac{n_i}{n}\right)^2 \left(\frac{1}{n^2}\right)$$

$$n^2 = \frac{\mu_p}{\mu_n} \cdot \left(\frac{n_i}{n}\right)^2$$

$$\boxed{n = n_i \sqrt{\frac{\mu_p}{\mu_n}}} \quad \text{--- (iv)}$$

Putting this value in eqⁿ (i)

$$\sigma = n_i q \sqrt{\mu_n \mu_p} + q\mu_p \cdot \frac{n_i^2}{n_i \sqrt{\frac{\mu_p}{\mu_n}}} \sqrt{\frac{\mu_p}{\mu_n}}$$

$$\sigma = n_i q \sqrt{\mu_n \mu_p} + q\mu_p n_i \sqrt{\mu_n \mu_p}$$

$$\boxed{\sigma = 2n_i q \sqrt{\mu_n \mu_p}} \quad \text{--- (v)}$$

eqⁿ (v) is the equation to calculate the minimum conductivity.

By eqⁿ (v) + eqⁿ (ii)

$$P = n_i \sqrt{\frac{\mu_n}{\mu_p}}$$

$$\sigma_{\min} = 2q n_i \sqrt{\mu_n \mu_p}$$

Question: If drift velocity of holes under a field gradient of 100 V/m is 5 m/sec find its mobility

Solution:

$$\mu = \frac{v}{E} = \frac{5}{100} = 0.05 \text{ cm}^2/\text{v}\cdot\text{sec}$$

Question: The carrier mobility in a semiconductor is 0.4 m²/v-s if diffusion constant at room temp^r is _____

$$D = \mu V_T \\ = 0.4 \times 0.026$$

$$D = 0.0184 \text{ m}^2/\text{sec}$$

The minority carrier lifetime & diffusion constant in a sc material are 100 μsec & 100 cm²/sec respectively the diffusion length of charge carrier is

Solution

$$L = \sqrt{D\tau} \\ = \sqrt{100 \times 10^{-4} \times 100}$$

$$L = 1 \text{ cm} \quad \text{Answer}$$

Question: A flat aluminium strip of a resistivity of $3.44 \times 10^{-8} \Omega \text{ m}$, a cross sectional area of $2 \times 10^{-4} \text{ m}^2$ & a length of 5 mm. is subject to a current flow of 50 mA. find voltage drop across the bar

Solution:

$$V = I \cdot R \\ = 50 \times 10^{-3} \times 3.44 \times 10^{-8} \times \frac{5 \times 10^{-3}}{2 \times 10^{-4}} \\ = 50 \times 3.44 \times 2.5 \times 10^{-4} \\ = 125 \times 3.44 \times 10^{-4} \text{ volt} \\ = 43 \text{ mV}$$

Question A sc wafer (having negligible thickness) is 0.5 mm thick. a potential of 100 mV is applied across the thickness

- (a) what is the drift velocity if mobility is 0.2 m²/v-sec
(b) How much time is required for an electron to move across the thickness

Solution:

$$(a) \quad E = \frac{V}{d} = \frac{100}{0.5} = 200 \text{ V/m}$$

$$v_d = \mu E = 0.2 \times 200 \\ = 40 \text{ m/sec}$$

$$(b) \quad t = \frac{0.5 \times 10^{-3}}{40} = 1.25 \times 10^{-5} \\ t = 12.5 \text{ } \mu\text{sec}$$

Question: A small concentration of minority carriers are injected into a homogeneous sc crystal at one point & having an electric field E

In N-type SC

$$N_A = 0$$

$$n = N_D + P \text{ or } n \approx N_D$$

$$n \approx N_D$$

* N_D is called donor concentration or the density of donor atoms & it represents the no. of pentavalent atoms added to the SC.

$$N_D = \text{Total no of atoms/cm}^3 \times \text{Impurity ratio}$$

* In N-type SC free e^- concentration is approximately equal to N_D (the density of donor atoms).

* In N-type SC, the current is predominantly dominated by the flow of electrons.

* The conductivity of entire SC is

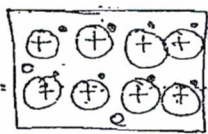
$$\sigma_N = nq\mu_n + pq\mu_p \text{ } \omega/\text{cm}.$$

$$\approx nq\mu_n \text{ } \omega/\text{cm}.$$

$$\sigma_N \approx N_D q \mu_n \text{ } \omega/\text{cm}.$$

* The conductivity due to minority carrier is almost negligible.

entire SC is represented as



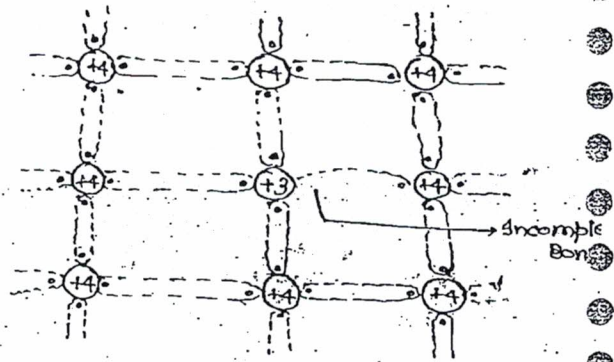
* (\oplus) ion \rightarrow it is a neutral atom with 5e less.

* ions are called immobile charge particles.

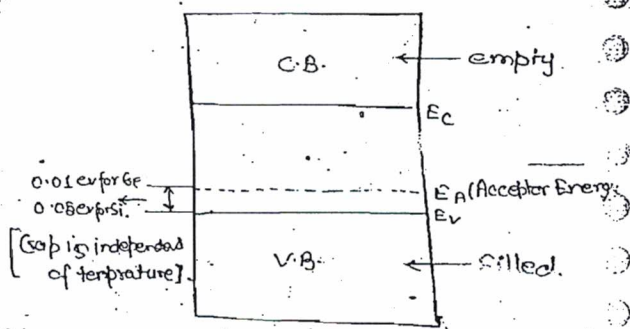
P-Type Semiconductor or Acceptors:-

* The impurity is trivalent.

Crystalline structure at 0°K -



Energy Band Diagram at 0°K



* Acceptor energy level is a discrete

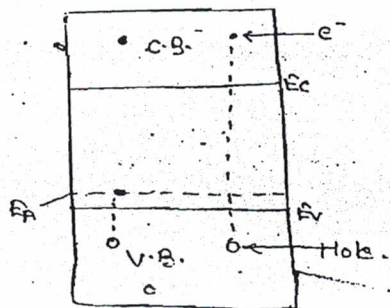
energy level created just above the valence band.

* Acceptor energy level denotes the energy level of trivalent atoms ad to the SC.

* P-type SC at 0°K will be working as an insulator.

Crystalline structure at 300°K





Energy Band Diagram at 300K:

* In p-type SC every impurity atom will be receiving $1e^-$ to complete its bonding hence p-type is also called as acceptor.

* $P \gg n$.

* Majority carriers are holes, minority carriers are e^- .

* when p-type SC is placed at room temperature, because of thermal energy a large no. of covalent bonds will be broken & equal no. of electron's & holes are generated most of these e^- 's will be moving into acceptor energy level to complete the bondings & very few e^- 's will be moving from valence band to conduction band so that the hole concentration in the VB is far greater than e^- concⁿ in the CB.

* In p-type SC current is mainly due to holes.

* The condition for p-type is

$$P \gg n \quad \text{or} \quad n \ll p$$

the e^- concⁿ is reducing below due to a large no. of bondings.

* according to the law of electrical neutrality -

$$N_D + P = N_A + n$$

Since $N_D = 0$.

$$P = N_A + n \quad \text{only for IES.}$$

or

$$P \approx N_A \quad \text{for all exams}$$

where N_A = acceptor concentration & it denotes the no. of trivalent atoms added to the SC.

$$N_A = \text{Total no. of atoms/cm}^3 \times \text{Impurity\%}$$

* The conductivity of p-type SC is

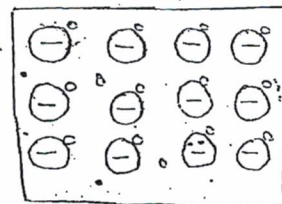
$$\sigma_p = n_e q \mu_n + p q \mu_p \quad \text{w/cm.}$$

$$\sigma_p \approx N_A q \mu_p \quad \text{w/cm.}$$

* The conductivity due to minority carriers is almost negligible.

* N-type SC is superior to P-type because $\mu_n > \mu_p$.

* p-type SC is represented as



LAW OF ELECTRICAL NEUTRALITY

* Based on law of conservation of charge

* Total positive charges = Total negative charges

$$N_D + P = N_A + n$$

N_D = donor concentration is associated with positive charge.

N_A = Acceptor concentration is associated with negative charge.

or

$$N_D - N_A = n - p$$

* A sc which obeys law of electrical neutrality is always electrically neutral.

for N-type sc

$$N_A = 0, n = N_D + p$$

$$n \approx N_D$$

for P-type sc

$$N_D = 0, p = N_A + n$$

$$p \approx N_A$$

In intrinsic sc

$$N_A = 0, N_D = 0$$

$$p = n$$

$$p - n = 0$$

∴ Intrinsic sc is electrically neutral.

∴ All sc are electrically neutral.

Question: N-type sc is _____

(a) (-)vely charged (b) (+)vely charged

(c) No charge at all (d) electrically neutral.

Question: A pure sc (Ge) is doped with donor impurities to the extent of $1:10^7$ calculate

(i) donor concⁿ

(ii) electron & holes concⁿ in the doped sc.

(iii) conductivity & resistivity of the doped sc.

* How many times the resistivity is increased

in the sc due to doping

assume total no. of atoms = $4.421 \times 10^{22} / \text{cm}^3$

$$n_i = 2.5 \times 10^{13} \text{ atoms/cm}^3$$

$$\mu_n = 3800 \text{ cm}^2/\text{v-sec} \quad \mu_p = 1800 \text{ cm}^2/\text{v-sec}$$

Solution:

$$N_D = 4.421 \times 10^{22} \times \frac{1}{10^7}$$

$$(i) N_D = 4.421 \times 10^{15} / \text{cm}^3$$

$$(ii) e^- \text{ conc}^n = N_D = 4.421 \times 10^{15} / \text{cm}^3$$

$$N_A = \frac{n_i^2}{N_D} = \frac{(2.5 \times 10^{13})^2}{4.421 \times 10^{15}}$$

$$N_A = 1.414 \times 10^{11} / \text{cm}^3$$

(iii)

$$\sigma = q N_D \mu_n$$

$$= 1.6 \times 10^{-19} \times 4.421 \times 10^{15} \times 3800$$

$$= 26.87968 \times 10^{-4}$$

$$= 2.688 \text{ } \Omega/\text{cm}$$

$$(iv) \rho_n = \frac{1}{\sigma_n} = 0.372 \text{ } \Omega \cdot \text{cm}$$

$$(v) \sigma_i = n_i q (\mu_n + \mu_p)$$

$$= 2.5 \times 10^{13} \times 1.6 \times 10^{-19} (3800 + 1800)$$

$$= 0.0224 \text{ } \Omega/\text{cm}$$

By adding donor impurities of $1:10^7$

σ ↑ from $0.0224 \text{ } \Omega/\text{cm}$ to $2.68 \text{ } \Omega/\text{cm}$

$$\uparrow \sigma \text{ by doping} = \frac{2.68}{0.0224} = 119.8$$

$$\approx 120 \text{ times}$$

Question: A pure sc (Si) is doped with

acceptor impurities to the extent of 4

impurity atoms / 10^6 atoms find its

conductivity Total no. of atoms = $8 \times 10^{22} / \text{cm}^3$